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Review Article

Silicon Nanowires with Mesopores: Fabrication and Optical Properties

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Abstract

Silicon nanowires (SiNWs) are electrochemically porosified to generate mesopores in sizes of 2-50 nm, resulting in the creation of mesoporous silicon nanowires (mpSiNWs). The porosification imposes two characteristics on to SiNWs: an increase of surface areas in the porous profile and quantum confinement ascribed to the shrinkage of silicon skeletons. Since the first report on producing mpSiNWs in 2009, the emerging porous nanostructures are increasingly catching fundamental and technical interests, and diverse but limited applications have been explored in batteries, energy storage, photocatalysis, drug delivery and gas sensors. In this mini review, we will elaborate the porosification mechanism in metal-assisted chemical etching of Si wafers that has been dominantly employed to fabricate mpSiWNs, in terms of intrinsic properties of Si wafers (i.e. the doping level and dopant elements), ingredients of the etching solution, temperature and complementary porosification methods. Among various physical properties of mpSiWNs, photoluminescence (PL) has been intensively studied owing to the promising application of mpSiNWs in optoelectronics. PL of mpSiNWs sensitively varies with the intrinsic properties of parent Si wafers, temperature and porosification processes, and the origin is still ambiguous and under debate. It will be reviewed the current studies in PL, together with the antireflection that is highly desired in producing green energy. At the end, it will be envisaged the prospects of mpSiNWs for developing the integrated optoelectronic devices.

ABBREVIATIONS

mpSiNWs: Mesoporous Silicon Nanowires; mpSi: Mesoporous Silicon Layers; MACE: Metal-Assisted Chemical Etching; 1-MACE: One-Step MACE; 2-MACE: Two-step MACE; NPs: Nanoparticles; SHE: Standard Hydrogen Electrode; \mathbf{n}'/\mathbf{p} -: n/p-lightly-doped (electrical resistivity ρ >0.1 Ω° cm); $\mathbf{n}^{+}/\mathbf{p}^{+:}$ n/p-medially-doped (0.005< ρ <0.1 Ω° cm); $\mathbf{n}^{++}/\mathbf{p}^{++}$ -:n/p-heavily-doped (ρ <0.005 Ω° cm); PL: Photoluminescence; QC: Quantum Confinement; RRCs: Radiative Recombination Centers; NRCs: Non-Radiative Recombination Centers

INTRODUCTION

SiNWs appear to be solid long cylinders without pores. It has been reported in 2009 that SiNWs were electrochemically porosified to create mpSiNWs with pores in sizes of 2-50 nm (*i.e.* mesopores) [1,2]. mpSiNWs integrate the one-dimensional nanostructures with two porosification-induced effects, so as to increasingly catch fundamental and technical interests recently. The two porosification-caused effects include an increase in

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surface area of the porous profile, and quantum confinement derived from the shrinkage in the residual Si skeletons. MACE has been predominantly employed to create mpSiNWs in a simple, low-cost way, and the fabrication mechanism will be fully reviewed. Compared to few reports on the properties of surface areas [1], electronic properties [1,2], surface hydrophobicity [3] and biodegradability [4], it has been intensively studied the optical properties of mpSiNWs, especially PL and antireflection. The porosification of SiNWs gives rise to various but limited applications in lithium ion batteries [5-7] energy storage [8], photocatalysis, [9] drug delivery, [10] and gas sensors [11]. These applications have been reviewed by Duan *et al.*, [12] and this mini review will mainly review the current studies in PL and antireflection of mpSiNWs.

Fabrication of mpSiNWs

mpSiNWs are produced by MACE of Si wafers, an anisotropic, galvanic etching [13] involving a series of (electro)chemical reactions, given by

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 $Ag^+ + e^- \rightarrow Ag(s)$ ($E^0 = 0.80$ V) (1)

$$Si(s) + 2H_2O - 4e^-?SiO_2(s) + 4H^+$$

(E°=-0.84 V)

$$SiO_2(s) + 6H \rightarrow H_2SiF_6 + 2H_2O$$

$$(\Delta H=-(138.22\pm 1.36) \text{ KJ/mol}) \tag{3}$$

(2)

 $(\Delta H = -(138.22 \pm 1.36) \text{ KJ/mol})$

$$H_2O_2 + 2H^+ + 2e^- ?2H_2O$$
 ($E^0 = 1.76$ V) (4)

$$2Ag(s) + H_2O_2 + 2H^+ \to 2Ag^+ + 2H_2O$$
 (5)

$$2Ag(s) + H_2O_2 + 2H^+ \rightarrow 2Ag^+ + 2H_2O$$

$$(E^o = 0 \text{ V})$$
(6)

The redox potential E is versus SHE. ΔH is the reaction enthalpy, and the negative sign illustrates that a reaction generates heat. MACE is classified as1-MACE and 2-MACE, which will be reviewed separately.

MACE of medially and heavily doped Si wafers: 1-MACE is generally operated in aqueous solution containing HF and AgNO₃ [14], involving the nucleation of Ag NPs and scratching (or sinking) of the growing NPs into the bulky Si wafers [15]. Ag⁺ ions in the vicinity of the wafer capture the electrons (eq. 1) that are donated by the anodic oxidation of Si (eq. 2), and then nucleate on the wafer surface. The Ag nucleation is thermodynamically ascribed to that Ag has electro negativity slightly higher than Si. HF dissolves the oxidized SiO₂ (eq. 3) to create pits that trap the Ag nuclei. Ag NPs preferentially adhere to surface states, such as dangling bonds, structural defects (steps, kink sites, and dislocations) and sites around the do pants [16]. The oxidation of one Si atom generates four electrons (eq. 2), and the reduction of an Ag⁺ ion consumes only one electron (eq. 1). The excess electrons donated from the wafer migrate to and accumulate on the surfaces of Ag NPs, and electrostatically drive Ag⁺ ions diffuse towards the nuclei for reduction. Owing to abundant substances in the electrolyte and wafer, eq. 1-3 can subsequently and repeatedly take place. As a result, Ag NPs continuously scratch and sink into the wafer, and simultaneously grow into dendrites. The residual of the scratched wafer appears to be an array of SiNWs without pores, which retain the intrinsic electronic properties of the parent wafer [17]. The scratching of Ag NPs is prohibited along the lateral direction perpendicular to the substrate normal, since the etching rate determined by the curvature of a pore reaches the minimum along the lateral direction [18]. The sinking direction of growing NPs governs the surface crystalline orientation of as-generated SiNWs, and is comprehensively determined by the crystalline structure of parent wafers, etching temperature and duration, and electrolyte concentration [19,20].

Solid SiNWs are usually created by 1-MACE of lightly and medially doped wafers and the porosification of SiNWs will be triggered by increasing the doping level of wafers. For instance, 1-MACE of n⁺⁺-Si(100) creates mpSiNWs [21]. The heavy doping-induced porosification can be understood by a thermodynamic model that describes charge transport across the silicon-electrolyte interface driven by potential energy (Figure 1a). Eq. 1 and 2 simply address the electron transport from Si to Ag⁺ ions, which is composed of the processes I-IV at the Si/electrolyte interface. Electrons at E_{ν} of Si wafers migrate to the interface by overcoming a potential energy barrier $\Delta \Phi$, consisting of the excitation barrier ($E_{\rm \tiny CB}\text{-}E_{\rm \tiny F'}$ the process I) and the band edge bending in SCL (E_{CBS} - $E_{CB'}$ the process II). Owing to the mismatch of $E_{{\it CB},{\it S}}$ with $E_{{\it Ag}}{}^{*},$ electrons are apt to relaxing to surface states (E_{ss}) aligned with E_{Ag}^{*} (the process III) followed by the resonant migration to E_{Ag}^{+} (the process IV). The increase in the doping level not only up shifts E_F towards E_{CB} to reduce $\Delta \Phi$, but also makes the SCL thinner to promote electron tunneling through the SCL (Figure 1b), to facilitate the process I and II. Hence, a large amount of Ag+ ions nucleate on the surface and [Ag⁺] (i.e. concentration of Ag⁺ ions) is significantly reduced in the electrolyte. Low [Ag⁺] can support only a portion of Ag nuclei to grow and scratch the bulk Si. The rest non-grown Ag nuclei will undergo a random scratching in a short period of time to porosify SiNWs and the bulky wafer (Figure 1c). Since the scratching of growing Ag NPs is limited by the diffusion of Ag⁺ ions with low [Ag⁺], the random porosification tends to be faster than the NP scratching. Consequently, the electrochemical porosification tends to generate a hybrid structure composed of an array of mpSiNWson mpSi (Figure 2a-d). On the contrary, the light and media doping increases $\Delta \Phi$ to prevent Ag⁺ ions from severely consuming. High [Ag⁺] substantially supports most Ag NPs grow and scratch the wafer to create SiNWs, and effectively suppresses the porosification (Figure 2e, f). This model can also account for the creation of mpSiNWs via 1-MACE of p**-Si [1].

To promote the porosification in 1-MACE of n⁺-Si, the process I and II are thermally excited via raising both the etching temperature from room temperature to 60 °C and [HF] from 2.4 mol/L to 5 mol/L [22]. The dissolution of SiO₂by HF is a heatreleasing reaction with a negative enthalpy (eq. 3), and the generated heat thermodynamically increases with [HF]. The thermal promotion of electron migration across the interfaces contributes to fabricating n⁺-mpSiNWs (Figure 3a) with the porosity gradually decreasing from the tops to roots (Figure 3bd), and there is a lack of mpSi underneath (Figure 3d). Compared to 1-MACE of n⁺⁺-Si, thermal excitation in 1-MACE of n⁺-Si generates less non-grown NPs, and thus the porosification of non-grown Ag NPs occurs slower than the scratching of growing Ag NPs, accounting for the generation of the gradient porous structures without mpSi (Figure 4).

Both the doping level and electro negativity of do pants play a significant role in the electrochemical etching. For example, the do pants of n-Si include P with electro negativity of 2.19, as of 2.18 and Sb of 2.05, and function as the surface states (E_{sc}). In 1-MACE of n⁺-Si, the doping with Sb can generate SiNWs, [21] but the Asdoping fails [2]. It is illuminated that the do pants are preferential to have small electro negativity for promoting the process IV. With respect to n⁺⁺-Si, a large amount of electrons accumulated in the vicinity of do pants overbalance the high electro negativity of As, so as to facilitate the process IV and create n⁺⁺-mpSiNWs via 1-MACE of As-doped n⁺⁺-Si [21]. In terms of n⁻-Si, surface defects mainly serve as the surface states instead of the do pants that have relatively low concentration, to decouple the do pant electro

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Figure 1 (a) Quasi-quantitative energy diagram at the silicon-electrolyte interface, at the beginning of 1-MACE of n⁺⁺-Si(100). The energy is referred to standard hydrogen electrode (SHE) potential. Heavily doped n-Si has Fermi energy (E_p), valence and conduction band in the bulk (E_{VB} and E_{CB}) and at the interface (E_{VBS} and E_{CBS}), and space charge layer (SCL) with a width W_{SCL} . $\Delta \Phi$ is an energy barrier between E_p and E_{CBS} . $E_{Ag}^{+}/Ag^{+}} E_{Ag}^{+}$ and E_{Ag} represent the potential energy of the redox pair, Ag⁺ and Ag in the electrolyte, respectively. λ is the reorientation energy. The light blue sphere represents an electron at E_p and the I-IV processes denote a subsequent series of electron migration from the bulk to interface. (b) Energy diagram with variation of doping levels in n-Si(100). (c) Schematic of the 1-MACE evolution. For clarity, it is shown only one non-grown Ag nucleus scratching the wafer in the mpSi, marked by a red dashed line. Within a period of time Δt , the evolution of the porosification and mp-SiNWs are denoted by ΔT_p and $\Delta T_{NW^{*}}$ respectively. (Adapted from Ref. [21]. Copyright 2011ACS Publications).



Figure 2 1-MACE of n^{++} -Si(100) generates mpSiNWs and a layer of porous Si (mpSi) underneath (a-d), and that of n^{-} -Si(100) (e) and n^{+} -Si(100) (f) creates SiNWs without mpSi. (a, b): TEM images; (c): SEM cross-sectional view; (d-f): SEM oblique views. (Adapted from Ref. [21]. Copyright 2011ACS Publications).



Figure 3 (a) A SEM image of n⁺-mpSiNWs. (b-d) High-magnification SEM images of the portion (I), (II) and (III) in (a). (Adapted from Ref. [22]. Copyright 2012ACS Publications).



Figure 4 (a) A SEM cross-sectional image of n⁺-mpSiNWs having the trapezoid structure. (b,c) Schematic illustrations of the generation of trapezoid n⁺-mpSiNWs via 2-MACE of n⁺-Si. (Adapted from Ref. [29]. Copyright 2011ACS Publications).

negativity with the generation of SiNWs. Therefore, SiNWs can be fabricated by MACE of P-doped n⁻-Si [17,21,23].

It is imperative to control the etching direction of growing Ag NPs for engineering surface crystalline orientation of mpSiNWs, [24] which plays a crucial role in catalysis, electronics and surface chemistry. The back-bond theory addresses that the etching along (100) need to cleave two Si-Si back bonds,

compared to three back bonds along (111) [16]. As a result, the etching is preferential along (100) over (111), which has been widely demonstrated in1-MACE of Si(100), Si(111) and Si(110) [17,19,20,25,26]. However, Huang *et al.* found that under [HF]/ [AgNO₃] \geq 50, 1-MACE of n⁻Si(111) has an intrinsic etching along [111] at room temperature, and the porosification in 1-MACE of n⁺⁺-Si(111) effectively disturbs the intrinsic etching [27]. The evolution of the porosification of non-grown NPs is faster than

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the scratching of growing NPs, to weaken the back bonds in front of the scratching NPs and deviate the etching from (111). The etching preferentially switches to a direction with a small angle to (111), since the direction-switching barrier increases with the angle. The etching is preferential along <211> at 10 °C, and <311> at 20 °C. The intrinsic etching along (111) is reserved above 30 °C, due to the thermodynamic preference to the intrinsic etching. It is illuminated that the etching is governed by not only the backbond etching, but also the crystalline structures of the as-etching wafers, concentration of etchants and etching temperature.

MACE of medially and heavily doped Si wafers: In 2-MACE, the nucleation of metal NPs and scratching of NPs are separately carried out [28]. The nucleation can be formed either by 1-MACE in a short period of time, or by the direct deposition of metal NPs in a random or periodic pattern. Then the etching is operated in aqueous HF solution containing strong oxidants (e.g. H_2O_2) instead of metallic ions. Metal NPs adhere on the surface of Si wafers and function as catalysts. Since $E^0_{H202/H20}$ is much more positive than $E^o_{_{SiO2/Si'}}$ H₂O₂ diffusing to the vicinity of adhered NPs stimulates the oxidation of Si underneath the NPs (eq. 2), and the oxides are dissolved by HF (eq. 3). The released electrons enter into the NPs and migrate to the NP-electrolyte interfaces, followed by reducing H_2O_2 (eq. 4). The subsequent repeated occurrence of (eq. 2-4) results in the scratching of adhered NPs to create SiNWs. If the catalysts NPs are made of Ag, $E^{0}_{_{H202/H20}}$ is also much more positive than $E^{0}_{_{Ag+/Ag}}$. Consequently, $H_{2}O_{2}$ can oxidize Ag NPs into Ag⁺ ions (**eq. 5**). Note that in the 2nd step of 2-MACE, there is no Ag+ ion in the etching solution. H₂O₂ with low concentration generates small amount of Ag⁺ ions in the vicinity of adhered Ag NPs, and the oxidized Ag⁺ ions most likely re-deposit onto the NPs. More Ag⁺ ions are generated with an increase of $[H_2O_2]$, and tend to diffuse away from the adhered NPs to the walls of as-generated SiNWs. Analogue to 1-MACE, the migrating Ag⁺ ions stimulate the electrochemical etching in the SiNWs' walls to roughen the surfaces of SiNWs at low $[H_2O_2]$, and to porosify SiNWs at high [H₂O₂] [29]. The porosity of mpSiNWs increases with $[H_2O_2]$ [2]. The H_2O_2 -induced dissolution of Ag NPs causes the catalyst NPs shrink during the etching (Figure 4b), to generate tapered mpSiNWs (Figure 4a,c) [29]. There is no mpSi underneath, since the scratching of shrunk Ag NPs is faster than the lateral porosification in the walls of SiNWs. The addition of surfactants (e.g. ethanol) in the 2nd-step etchant solution (HF and H_2O_2) can effectively slow the scratching of Ag NPs, and thus generate a mpSi underneath mpSiNWs, even though the reason is ambiguous [4]. The diffusing Ag⁺ ions that are oxidized by H_2O_2 preferentially nucleate on surface states nearby the do pants. The more do pants, the higher porosity SiNWs will have. Therefore, mpSiNWs are generally produced via 2-MACE of medially or heavily doped Si wafers. It should be emphasized that the porosification of SiNWs is attributed to the H₂O₂-induced oxidation of Ag NPs. To verify the role of H_2O_2 , H_2O_2 was replaced with Fe(NO₃)₃ [30]. Since $E^{0}_{Fe3+/Fe2+}$ is slightly more negative than $E^{0}_{Ag+/Ag}$, Fe³⁺ ions thermodynamically fail to oxidize the adhered Ag NPs, so as to create SiNWs instead of mpSiNWs [31].

At $[H_2O_2]$ of 0.5 mol/L, the porosity of mpSiNWs increases with [HF]. When [HF] is increased to 5.8 mol/L, the HF-induced dissolution of SiO₂ is quick enough to remove the apexes of mpSiNWs, resulting in the formation of conical mpSiNWs [32]. When the nucleation is periodically patterned on Si wafers, 2-MACE can generate periodic arrays of mpSiNWs with controllable diameters and lengths. The template method includes nanosphere lithography, [33-35] anodic aluminum oxide mask method, [24] interference lithography, [36,37] superionic solid state stamping, [23] and block-copolymer mask method [38].

Generation of mpSiNWs by MACE of lightly doped Si wafers: Using both 1- and 2-MACE, mpSiNWs are generated from medially and heavily doped wafers. The light doping creates a high $\Delta \Phi$ to suppress the electron transport process I and II, and to eliminate the porosification in SiNWs. To overcome the big $\Delta \Phi$, after the generation of n-SiNWs via 1-MACE of n-Si the wafer was further etched at a bias of 45 V and current density of 7 mA/cm² [3]. The electrochemical cell was composed of an anodic electrode of the n-Si wafer with n-SiNWs, and a cathodic electrode of a bare n⁻-Si wafer. The electrolyte was an aqueous solution containing 4 mol/L HF and 3.2 mol/L H_2O_2 . On the anodic electrode, the extremely high bias effectively drives the oxidation of Si (eq. 2) to cause the lateral porosification in SiNWs via eq. 3. The as-generated H⁺ ions (via eq. 2) electrically migrate to the cathode for reducing H₂O₂ (eq. 4). The porosity of n⁻-mpSiNWs increases with elongating the duration of the high-bias etching.

Another method of generating lightly doped mpSiNWs is 1-MACE of lightly doped Si wafers in the etching solution containing 4.6 mol/L HF, 0.02 mol/LAgNO₃ and H_2O_2 [39]. During 1-MACE, the scratching Ag NPs grow into a dendritic Ag layer to cover the wafer. The addition of H_2O_2 generates a large amount of Ag⁺ ions by oxidizing Ag dendrites, resulting in the lateral porosification in the smooth walls of as-generated SiNWs. Low $[H_2O_2]$ of 0.005 mol/L generates small amount of Ag⁺ ions, and only roughens SiNWs without porosification. mpSiNWs are created at $[H_2O_2]$ increasing to 0.05 mol/L. The etching at >60 °C promotes the H_2O_2 -induced oxidation of Ag dendrites and Ag⁺ ion diffusion, to facilitate the fabrication of mpSiNWs [40]. This method is adapted to n⁻Si and p⁻Si.

Photoluminescence of mpSiNWs: PL originates from the radiative recombination of excitons, which are optically generated by laser. The emission wavelength of PL is determined by the energy difference of two density of states between which the radiative recombination occurs [41]. The assignment of PL lies in identifying the nature of density of states, in which the recombining excitons stay. PL characteristics of mpSiNWs (i.e. the intensity, peak position λ_{pl} and decay time) are very sensitive to the doping level, [42] porosity and dangling bonds at the surfaces [43], Yang et al., reported that in the ambient condition, p⁺⁺-mpSiNWs have a broad, roughly symmetric PL peak centered at 680 nm (Figure 5a). They simply attributed the reddish PL to deep QC in the shrunk Si skeletons of p**-mpSiNWs, i.e. the direct radiative recombination across the Si band gap which is widened and transformed from indirect to direct due to the porosificationinduced QC [1]. It is derived from QC that $\lambda_{_{PL}}$ of 680 nm requires to reduce the size of Si skeletons to3-4 nm, which was verified by TEM of individual p⁺⁺-mpSiNWs (Figure 5b). The broad peak is ascribed to wide distribution in the size of Si skeletons. The broad, symmetric reddish PL has been observed in n-mpSiNWs, [3] n⁺-mpSiNWs, [2,43] p⁺-mpSiNWs, [44] and n⁺⁺-mpSiNWs [21].



Figure 5 (a) PL spectrum of p⁺⁺-mpSiNWs on p⁺⁺-Si. (b) A highmagnification TEM image of a p⁺⁺-mpSiNW. (Adapted from Ref. [1]. Copyright 2009ACS Publications).

However, it is under debate that the reddish PL originates from QC. First, QC predicts a blue shift of PL with increasing the porosity of mpSiNWs. Chiappini et al., increased the porosity of p⁺⁺-mpSiNWs by raising [H₂O₂]. They found that low porosity have a broad PL spectrum centered at ~450 nm, but high porosity causes a red shift of λ_{p} to ~540 nm [4]. Bai *et al.*, used 1-MACE to generate p⁻ and p⁺⁺-mpSiNWs. Compared to p⁺⁺-mpSiNWs with high porosity, p-mpSiNWs have porosity reducing from the sides to cores. p-mpSiNWs emit green PL, and p++-mpSiNWs have reddish PL] [39]. The red shift with porosity doesn't obey QC. Second, the oxidation of mpSiNWs shrinks Si skeletons and should make PL blue shift, according to QC. Focused laser with power larger than 55 mW was employed to oxidize n⁺⁺-mpSiNWs. The laser-induced oxidation effectively reduces the porosity (Figure 6a versus 6b). However, the reddish PL tends to slightly red shift from \sim 750 nm to \sim 775 nm (Figure 6c, as marked by the arrows) [45]. Third, electrochemical etching was used to fabricate nmpSiNWs that have a broad reddish PL with $\lambda_{_{PL}}$ of 680 nm. It was from the TO (transverse optical)-phonon Raman mode located at 520.7 cm⁻¹that the average size of silicon skeletons was evaluated as 6.6 nm [3]. QC substantially requires to have a characteristic size smaller than the Bohr radius of the free excitons of bulky Si (~5 nm), but the Si skeletons were evaluated larger than 5 nm [46]. Fourth, HF removes the oxide layers of n⁺-mpSiNWs and significantly causes λ_{ν} red shift from ~700 nm to ~800 nm [47], However, Si is intact in HF, so it is indicated by QC that the reddish PL not shift after the HF etching. The removal of oxides does affect



Figure 6 TEM images: (a) pristine n⁺⁺-mpSiNWs, (b) laser-treated n⁺⁺-mpSiNWs. (c) PL spectrum of the pristine and laser-treated n⁺⁺-mpSiNWs. (d) PL spectrum of laser-treated n⁺⁺-mpSiNWs as function of laser power. (e) A plot of the integrated PL intensity (in the range of 400-600nm) versus laser power, divided into the region I, II, IIIA and IIIB. (Adapted from Ref. [45]. Copyright 2014 Nature Publishing Group).

the reddish PL, indicating that the reddish PL relate to surface states at the oxide/Si interfaces. The electrochemically-induced porosification of n⁻SiNWs significantly suppresses the vibrations of interfacial Si-O-Si, O-SiH_x and SiH_x, illustrating that the reddish PL should not be ascribed to hydrogenated amorphous Si, surface hydrides or polymer molecules on the surfaces [3]. It was proposed that surface states are the defects in the oxide layers [47]. XPS detects Si⁰ (Si), Si⁺ (Si₂O), Si³⁺ (Si₂O₃) and Si⁴⁺ (SiO₂) in n⁺⁻mpSiNWs. The surface defects could be Si⁺ (Si₂O) and Si³⁺ (Si₂O₃) that are generated by dangling bonds and volumetric stress at the oxide/Si interfaces during the porosification [32].

Investigation in the dependence on temperature can shed light on the origin of the reddish PL of mpSiNWs covered with native oxides. p⁺-mpSiNWs emit reddish PL with λ_{PL} of ~650 nm at 300 K (Figure 7a), which red shifts from ~590 nm at 15 K to 650 nm at room temperature (Figure 7b). When the excitons migrate from RRCs to NRCs over an energy barrier, the excitons will undergo a non-radiative recombination to quench PL. The migration can be thermally promoted to overcome the RRCsto-NRCs energy barrier, so the quenching is so-called thermal quenching. Thermal quenching accounts for that the PL intensity decreases gradually with temperature in the range of 12-160 K and 220-300 K (Figure 7c). There is a little increase in the range of 160-220 K, and a multi-level transition model was proposed to account for the negative thermal quenching [44]. In the ambient conditions, n⁺⁺-mpSiNWs emit reddish PL centered at ~700 nm (Figure 7d), but have different dependence on temperature (Figure 7e) [21], Figure 7f shows that $\lambda_{_{PL}}$ tends to have small fluctuation in the range of 16-220 K, followed by a red shift with temperature in the range of 220-290 K. The intensity gradually increases with temperature in the range of 16-160 K, ascribed to the thermo activated anti-trapping from surface defects in RRCs. Then the intensity diminishes with temperature in the range of 160-290 K, owing to thermal quenching. The reddish PL quenches quickly with temperature in the range of 160-220 K ascribed to high thermal quenching barrier of 150.6 meV, and reduces much more slowly in the range of 220-290 K attributed to small energy barrier of 31.2 meV. The different temperature dependence between p⁺- and n⁺⁺-mpSiNWs may be ascribed to the diverse interfacial states, doping level and do pant elements.

The reddish PL of n⁺-mpSiNWs (Figure 8a) has λ_{p_l} insensitive to temperature, excitation laser energy (Figure 8b) and laser power. But the PL intensity has a monotonous decrease with temperature in the range of 100-300 K (Figure 8c), increases with laser energy, and reaches the maximum at laser power of 0.35 mW [48]. When the oxides are removed by HF, the H-terminated n*-mpSiNWs emit a broad, asymmetric PL spectrum decomposed into two peaks: a reddish peak centered at ~730 nm and NIR peak (Figure 8d). The reddish and NIR PL both have the maximum intensity at 160 K (Figure 8e and 8f), but have different dependence of λ_{PL} on temperature and laser energy. λ_{PL} of the reddish PL is insensitive to temperature and laser energy, but the NIR PL blue shifts with temperature (Figure 8f) and laser energy. It was proposed that the triplet (T)-to-singlet (S) state transition dominates below 160 K. Owing to the exchange interaction of electrons and holes, the excitonic levels could split



Figure 7 (a) PL spectrum of p⁺-mpSiNWs at room temperature. Inset is the photograph of the p⁺-mpSiNW array under the illumination of He-Cd 325 nm laser. (b) Temperature-dependent PL spectra of p⁺-mpSiNWs. (c) Temperature-dependent PL peak intensity. (Adapted from Ref. [44]. Copyright 2011 AIP) (d) PL spectrum of n⁺⁺-mpSiNWs at room temperature. (e) Temperature-dependent PL spectra of n⁺⁺-mpSiNWs. (f) The plots of peak intensity and peak center versus temperature. (Adapted from Ref. [21]. Copyright 2011ACS Publications).



Figure 8 (a) A TEM image of n⁺-mpSiNW. (b) Excitation energy-dependent PL spectra of n⁺-mpSiNW. (c) Temperature-dependent PL intensity. (d) Temperature-dependent PL spectra from 80 K to 290 K. (e) Temperature-dependent PL intensity for the red PL fixed at 730 nm. (f) Temperature-dependent PL intensity and peak wavelength of NIR PL. (Adapted from Ref. [48]. Copyright 2011 AIP).



Figure 9 (a) Reflective spectra of n⁻-SiNWs and n⁻-mpSiNWs. n⁻-mpSiNWs were created by electrochemically porosifying n⁻-SiNWs as function of porosification duration in the range of 1-3 min. (b) The zoom-in of (a) in the low reflective spectra. (c) A TEM image of a n⁻-mpSiNW. (Adapted from Ref. [3]. Copyright 2012 Elsevier).

into a dipole-allowed upper singlet level and a dipole-forbidden lower triplet level. At low temperature, the excitons mainly occupy at the energetic favorable triplet state. The probability of T-to-S transition becomes higher with increasing temperature, leading to a raise of PL intensity. Above 160 K, thermal quenching tends to dominate over the T-to-S transition. The temperaturedependent tradeoff between the T-to-S transition and thermal quenching makes the reddish and NIR PL intensity reach the maximum at 160 K. The T-to-S transition happens in the H-terminated n⁺-mpSiNWs. The transition-induced excitons can migrate from Si skeletons to interfacial states with smaller sizes nearby. The difference in the dependence of $\lambda_{_{PL}}$ on temperature and laser energy illustrates that the reddish and NIR PL originate from two different kinds of interfacial states. When n*-mpSiNWs are covered with native oxides, the surface oxides prevent the T-to-S transition, resulting in the domination of thermal quenching in the range of 80-290 K (Figure 8c). In consistence, Lin et al., reported that the PL spectrum of H-terminated n⁺mpSiNWs is decomposed into a reddish peak at ~750 nm and NIR peak at \sim 850 nm [49]. The NIR PL enhances in intensity with the porosity faster than the reddish PL, leading to the red shift of the total broad PL spectrum with the porosity.

PL can be engineered by surface modification of mpSiNWs. Choi *et al.*, operated focused green laser to fast scan an array of n^{++} -mpSiNWs, and additionally generated a broad greenish-blue PL spectrum centered at ~550 nm (Figure 6c) [45]. The threshold to excite the greenish-blue PL is the laser power of ~5 mW, and the intensity increases with laser power in the range of 5-105 mW, followed by a quenching beyond 105 mW (Figure 6d-e). As characterized by Raman spectrum, in the region II (5-55 mW) the greenish-blue PL results from surface states generated by the laser-induced quick melting/re-solidification on the surfaces of n^{++} -mpSiNWs. Fast laser scanning causes surface oxidation in the region III (>55 mW). In the region IIIA (55-105 mW), the greenish-blue PL is attributed to the oxide defects and oxygen holes in the oxide layers. The oxygen holes constructively couples with Si phonons to elongate the phonon lifetime, but the oxide

defects reduce the phonon lifetime by destructive interference. Beyond 105 mW (the region IIIB), the laser-induced oxide layers turn to be stoichiometric, resulting in a diminishment in the interfacial concentration of the oxygen holes and the dominant contribution of the oxide defects. The laser-induced up shift of density of states of the oxide defects causes a decrease in density of states in the greenish-blue band, accounting for the quenching beyond 105 mW. The greenish-blue PL can be enhanced by the post-laser annealing at 300 °C, in vacuum of 10⁻² Torr.

When p-mpSiNWs are conformally coated with 6-10 nm polycrystalline SnO₂ by atomic layer deposition, PL appears to have a slight red shift from 510 to 530 nm [50]. By annealing the SnO₂-coated p⁻-mpSiNWs in N₂/H₂ at 600 °C, SnO₂ is chemically decomposed into Sn, which diffuses into the Si skeletons to enhance PL. Zhang et al, .fabricated n*-mpSiNWs to emit a broad, symmetric PL spectrum centered at ~725 nm. n⁺-mpSiNWs were selenized at 700 °C in vacuum of 4×10⁻⁴ Pa [43,51]. FTIR shows the Si-Se bonds replace Si-H bonds on the surfaces, to passivate the dangling bonds and enhance PL stability during aging. XPS detects Si-Se at 54.9 eV and Si-Se-O at 56.3 eV, illustrating that sufficiently diffuses into n⁺-mpSiNWs. When exposed to a 532-nm laser, the selenized n⁺-mpSiNWs generate a broad PL spectrum decomposed into three peaks centered at 595, 645 and 715 nm. The peak centered at 595 nm and 645 nm are ascribed to the Si skeletons passivated by Si-Se-O and Si-Se, respectively. The third peak is retained from the pristine n⁺-mpSiNWs. The blue shift of the laser from 532 to 330 nm eliminates the reddish PL at 715 nm, and the two Se-related PL peaks blue shift from 595 to 510 nm and from 645 to 610 nm, respectively. The time-resolved PL spectrum was monitored to evaluate that PL stemming from Si-Se-O and Si-Se has an exciton lifetime of 2.68 and 0.49 ns, respectively. As comparison, the pristine n⁺-mpSiNWs emit the reddish PL at 715 nm with lifetime of 54 µs. The significant reduce in the lifetime induced by the selenization accounts for the selenization-caused enhancement of PL. Furthermore, the selenization generates a broad NIR PL spectrum centered at ~1300 nm, probably attributed to the recombination of excitons in defect states.

Antireflection of mpSiNWs

MACE generates an array of vertical mpSiNWs on a Si wafer, which functions as an antireflection coating to effectively reduce reflection loss of smooth wafers and magnify photovoltaic efficiency of Si-based solar cells [52]. The contribution of antireflection lies in two factors. First, incident light will be multiply reflected among close-packed vertical NWs to prevent light from escaping, i.e. optical trapping [53]. Second, an antireflection coating has an index of refraction to reduce the mismatch between the air and Si wafer. It is highly desirable for antireflection to have a tapered structure, in which the volume fraction of Si gradually reduces from the wafer surface to the air [54]. Compared to high reflectance of bare Si wafers in the range of 80%-36% in the UV-visible spectrum, n-SiNWs reduce reflectance to 2.5-5.5% mainly due to light trapping (Figure 9a) [3]. The electrochemical porosification of n-SiNWs can further reduce reflectance with increasing the porosity (Figure 9b), owing to the decrease in refraction index and increase in surface roughness (Figure 9c). However, To et al., used an integrated sphere to measure scattering and transmittance, finding that an array of n^{**} -mpSiNW strap less light and scatter more light than n**-SiNWs with the same thickness [53]. The porosification-enhanced light scattering may originate from the surface roughening induced by the porosification. Therefore, the enhanced antireflection with the porosity of n⁻-mpSiNWs (Figure 9b) is more likely attributed to the tapered structures (Figure 9c). Recently, Charrier et al., created n*-mpSiNWs that have not only the tapered profile (Figure 10a), but also the porosity gradually reducing from the top to root portions (Figure 10b) [47]. The gradient-porosity structures give rise to a graded decrease of refraction index from the Si wafer to the air, resulting in specular reflectance less than 0.1% in the UV-visible spectrum at an incident angle of 6° (Figure 10b). The elongation of tapered n⁺-mpSiNWs can further reduce reflection loss, owing to the enhancement of optical trapping (Figure 10c).

Prospects

Optoelectronic devices substantially require an integration of



Figure 10 TEM images: (a) a n^+ -mpSiNW with the cone shape, (b) two n^+ -mpSiNWs having graded porosity along their longitudinal axis. (c) Experimental and theoretical reflectance spectra of n^+ -mpSiNWs as function of the NW length. (Adapted from Ref. [47]. Copyright 2013Elsevier).

tailorable electronic and optical properties of one-dimensional nanostructures, and there are some prospective outlooks for the investigation and development of mpSiNWs. First, It should be further studied charge transport in mpSiNWs, how much it deviates from parent Si wafers, the gate effect, and the dependence on the porosity. Second, surface modification plays an important role in enhancing anti-aging, passivating surface defects, tuning surface hydrophobicity, controlling exciton recombination, engineering charge transport, and improving chemical sensing. The grafting of SiNWs has been reported, including the silanization of SiO_covered SiNWs, hydrosilylation of H-terminated SiNWs, and twostep chlorination/alkylation of H-terminated SiNWs [55]. It should be explored the feasibility to impose these grafting approaches on to mpSiNWs. Third, the origin of PL is still ambiguous. Because PL is closely related to considerable amount of surface states/ defects created by the porosification, the laser-induced doping of mpSiNWs in controlled environment (e.g. in H₂, Ar, N₂, organic solvents and polar solvents) could shed more light on the PL mechanism. Fourth, to make mpSiNWs adapted to a wide range of applications, it is highly desired the transfer of mpSiNWs onto a flexible, transparent substrate. Zheng et al., generated a layer of cracks in an array of SiNWs by partially porosifying SiNWs [56]. The mechanical breakdown preferentially takes place at the crack layer, and thus a portion of SiNW arrays can be transferred to an adhesive. They also generated a sacrificial mpSi underneath the array of SiNWs by electro etching the parent Si wafers. The sacrificial mpSi significantly reduces the adhesion of SiNWs to the wafer, to facilitate the transfer to other substrates [57]. Analogously, 1-MACE directly creates a sacrificial mpSi beneath an array of n⁺⁺-mpSiNWs (Figure 2c-d), making it highly possible to transfer. For those without a sacrificial mpSi underneath, the porosification of parent Si wafers can be carried out by the electrochemical etching to create mpSi.

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